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2,2-Dimethyl-5-[(5-methylfuran-2-yl)-methylidene]-1,3-dioxane-4,6-dione

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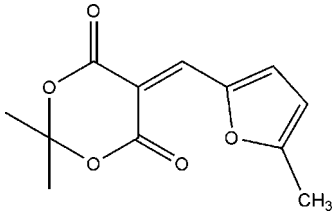
Key indicators: single-crystal X-ray study; T = 293 K; mean σ(C–C) = 0.003 Å; R factor = 0.044; wR factor = 0.156; data-to-parameter ratio = 17.4.

The asymmetric unit of the title compound, $C_{12}H_{12}O_5$, contains two independent molecules. In each, the 1,3-dioxane ring adopts an envelope conformation with the dimethyl-substituted C atom forming the flap. The crystal structure is stabilized by weak intermolecular C—H···O hydrogen bonds.

Related literature

For related structures, see: Zeng (2010*a*,*b*).



Experimental

Crystal data

$C_{12}H_{12}O_5$
M_r = 236.22
Triclinic, $P\overline{1}$
a = 8.9590 (18) Å
b = 10.038 (2) Å
c = 13.616 (3) Å
α = 92.71 (3)°
β = 105.99 (3)°
γ = 91.67 (3)°
V = 1174.7 (4) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.11 mm^{-1}
T = 293 K
0.20 × 0.16 × 0.12 mm

Data collection

Bruker SMART CCD area-detector diffractometer
11592 measured reflections
5336 independent reflections
3333 reflections with $I > 2\sigma(I)$
R_{int} = 0.024

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.044
$wR(F^2)$ = 0.156
S = 1.07
5336 reflections
307 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.31 e Å^{-3}
$\Delta\rho_{min}$ = −0.30 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
C1B—H1BC···O3B^{i}	0.96	2.48	3.434 (3)	170
C10B—H10A···O4A^{ii}	0.93	2.60	3.448 (3)	153
C10A—H10B···O1A^{iii}	0.93	2.44	3.343 (2)	163
C2A—H2AB···O2A^{iv}	0.96	2.57	3.524 (3)	170
C1A—H1AA···O3B	0.96	2.55	3.496 (3)	170

Symmetry codes: (i) $-x, -y+1, -z+1$; (ii) $-x+1, -y+2, -z+2$; (iii) $x, y-1, z$; (iv) $-x-1, -y+2, -z+1$.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH5196).

References

Bruker (1997). *SMART* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Zeng, W.-L. (2010*a*). *Acta Cryst.* E**66**, o2366.
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supplementary materials

Acta Cryst. (2011). E**67**, o478 [**doi:10.1107/S1600536811002285**]

2,2-Dimethyl-5-[(5-methylfuran-2-yl)methylidene]-1,3-dioxane-4,6-dione

W.-L. Zeng

Comment

In previous papers, the author recently reported the crystal structure of 5-(4-fluorobenzylidene)-2,2-dimethyl-1,3-dioxane-4,6-dione and (E)-2,2-dimethyl-5-(3-phenylallylidene)-1,3-dioxane-4,6- dione (Zeng, 2010*a*,*b*). As part of this search for new Meldrum's acid compounds, the title compound, (I) (Fig. 1), was synthesized and its crystal structure is reported herein. There are two symmetry-independent molecules, A and B, in the asymmetric unit of (I). The corresponding bond lengths and angles for the indepenent molecules agree well with each other as is reflected especially for the C7═C5 distance where the value and the standard uncertainty are the same in each. The 1,3-dioxane rings in each molecule are in envelope conformations. The crystal structure is stabilized by weak intermolecular C—H···O hydrogen bonds (Table 1).

Experimental

The mixture of malonic acid (6.24 g, 0.06 mol) and acetic anhydride(9 ml) in conc. sulfuric acid (0.25 ml) was stirred with water at 303K, After dissolving, propan-2-one (3.48 g, 0.06 mol) was added dropwise into solution for 1 h. The reaction was allowed to proceed for 2 h. The mixture was cooled and filtered, and then an ethanol solution of 5-methylfuran-2-carbaldehyde (6.60g,0.06 mol) was added. The solution was then filtered and concentrated. Single crystals were obtained by evaporation of an petroleum ether-acetone (3:1 *v*/*v*) solution of (I) at room temperature over a period of several days.

Refinement

The H atoms were placed in calculated positions (C—H = 0.93–0.96 Å), and refined as riding with U_{iso}(H) = 1.2U_{eq}(C) or 1.5U_{eq}(methyl C).

Figures

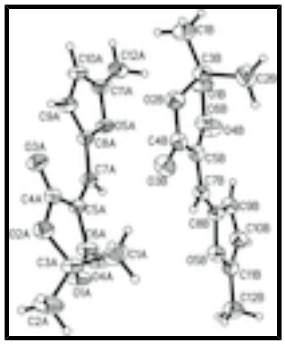


Fig. 1. The molecular structure of (I), drawn with 30% probability ellipsoids and spheres of arbritrary size for the H atoms.

2,2-dimethyl-5-[(5-methylfuran-2-yl)methylidene]-1,3-dioxane-4,6-dione

Crystal data

$C_{12}H_{12}O_5$	$Z = 4$
$M_r = 236.22$	$F(000) = 496$
Triclinic, $P\overline{1}$	$D_x = 1.336$ Mg m^{-3}
Hall symbol: -P 1	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
$a = 8.9590$ (18) Å	Cell parameters from 3333 reflections
$b = 10.038$ (2) Å	$\theta = 3.1$–$27.5°$
$c = 13.616$ (3) Å	$\mu = 0.11$ mm^{-1}
$\alpha = 92.71$ (3)°	$T = 293$ K
$\beta = 105.99$ (3)°	Block, yellow
$\gamma = 91.67$ (3)°	0.20 × 0.16 × 0.12 mm
$V = 1174.7$ (4) Å^3	

Data collection

Bruker SMART CCD area-detector diffractometer	3333 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube	$R_{int} = 0.024$
graphite	$\theta_{max} = 27.5°$, $\theta_{min} = 3.1°$
φ and ω scans	$h = -11 \rightarrow 10$
11592 measured reflections	$k = -13 \rightarrow 13$
5336 independent reflections	$l = -17 \rightarrow 17$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.044$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.156$	H-atom parameters constrained
$S = 1.07$	$w = 1/[\sigma^2(F_o^2) + (0.0907P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$
5336 reflections	$(\Delta/\sigma)_{max} = 0.001$
307 parameters	$\Delta\rho_{max} = 0.31$ e Å^{-3}
0 restraints	$\Delta\rho_{min} = -0.30$ e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
O2B	0.11969 (13)	0.55646 (11)	0.62210 (9)	0.0618 (3)
O5B	0.45369 (14)	1.05664 (10)	0.82774 (9)	0.0563 (3)
O1B	0.34137 (15)	0.48325 (11)	0.74003 (9)	0.0645 (3)
C5B	0.29935 (18)	0.71777 (15)	0.73236 (11)	0.0467 (4)
O3B	0.06239 (14)	0.76729 (12)	0.61279 (9)	0.0671 (3)
O4B	0.48827 (17)	0.61623 (12)	0.86197 (9)	0.0749 (4)
C6B	0.3859 (2)	0.60755 (16)	0.78319 (12)	0.0536 (4)
C7B	0.33715 (18)	0.84921 (15)	0.75839 (11)	0.0485 (4)
H7BA	0.2651	0.9054	0.7206	0.058*
C4B	0.15441 (19)	0.68642 (16)	0.65155 (12)	0.0522 (4)
C8B	0.46201 (19)	0.91860 (14)	0.83032 (11)	0.0481 (4)
C9B	0.5954 (2)	0.89135 (17)	0.90245 (12)	0.0564 (4)
H9BA	0.6302	0.8070	0.9202	0.068*
C3B	0.2442 (2)	0.46641 (16)	0.63735 (13)	0.0573 (4)
C11B	0.5813 (2)	1.11096 (17)	0.89765 (13)	0.0569 (4)
C10B	0.6699 (2)	1.01280 (18)	0.94451 (14)	0.0622 (4)
H10A	0.7635	1.0245	0.9955	0.075*
C1B	0.1707 (3)	0.32824 (18)	0.62731 (17)	0.0801 (6)
H1BA	0.1141	0.3201	0.6774	0.120*
H1BB	0.2501	0.2640	0.6381	0.120*
H1BC	0.1008	0.3123	0.5600	0.120*
C2B	0.3372 (3)	0.4908 (2)	0.56325 (16)	0.0780 (6)
H2BA	0.3813	0.5805	0.5749	0.117*
H2BB	0.2708	0.4787	0.4946	0.117*
H2BC	0.4191	0.4290	0.5728	0.117*
C12B	0.5961 (3)	1.25844 (18)	0.90908 (17)	0.0785 (6)
H12A	0.5088	1.2952	0.8619	0.118*
H12B	0.6903	1.2885	0.8950	0.118*
H12C	0.5986	1.2874	0.9778	0.118*
O5A	0.07339 (13)	0.59957 (9)	0.88357 (8)	0.0527 (3)
O2A	−0.36509 (13)	0.95250 (10)	0.63621 (9)	0.0573 (3)
O3A	−0.35857 (14)	0.73997 (11)	0.66614 (9)	0.0617 (3)
C5A	−0.14448 (18)	0.87879 (13)	0.76199 (11)	0.0447 (3)
O1A	−0.17387 (16)	1.11015 (10)	0.72684 (10)	0.0698 (4)
C8A	−0.05825 (18)	0.64850 (14)	0.81836 (11)	0.0447 (3)
C7A	−0.05103 (18)	0.78842 (14)	0.81679 (12)	0.0468 (4)
H7AA	0.0371	0.8277	0.8635	0.056*
C4A	−0.29180 (18)	0.84726 (14)	0.68578 (11)	0.0461 (4)
C6A	−0.0910 (2)	1.01900 (15)	0.78701 (14)	0.0596 (4)
O4A	0.01460 (18)	1.05984 (12)	0.85763 (12)	0.0924 (5)

C11A	0.0546 (2)	0.46486 (14)	0.87915 (13)	0.0529 (4)
C9A	−0.1573 (2)	0.54271 (14)	0.77519 (13)	0.0556 (4)
H9AA	−0.2548	0.5464	0.7287	0.067*
C3A	−0.2727 (2)	1.06729 (15)	0.62802 (13)	0.0592 (4)
C10A	−0.0843 (2)	0.42757 (15)	0.81423 (14)	0.0614 (5)
H10B	−0.1249	0.3404	0.7981	0.074*
C2A	−0.3843 (3)	1.17833 (18)	0.59790 (17)	0.0892 (7)
H2AA	−0.4385	1.1928	0.6490	0.134*
H2AB	−0.4578	1.1538	0.5332	0.134*
H2AC	−0.3273	1.2588	0.5924	0.134*
C12A	0.1835 (2)	0.39080 (17)	0.94262 (15)	0.0713 (5)
H12D	0.1569	0.2968	0.9333	0.107*
H12E	0.2009	0.4190	1.0134	0.107*
H12F	0.2763	0.4085	0.9223	0.107*
C1A	−0.1792 (3)	1.0368 (2)	0.55522 (18)	0.0901 (7)
H1AA	−0.1115	0.9659	0.5798	0.135*
H1AB	−0.1182	1.1149	0.5499	0.135*
H1AC	−0.2475	1.0099	0.4891	0.135*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O2B	0.0499 (7)	0.0631 (7)	0.0632 (7)	−0.0012 (5)	0.0031 (6)	−0.0101 (6)
O5B	0.0605 (7)	0.0498 (6)	0.0555 (6)	0.0040 (5)	0.0110 (6)	0.0001 (5)
O1B	0.0751 (9)	0.0494 (6)	0.0566 (7)	0.0003 (5)	−0.0020 (6)	0.0017 (5)
C5B	0.0457 (9)	0.0530 (8)	0.0388 (8)	0.0032 (6)	0.0071 (7)	0.0021 (6)
O3B	0.0566 (8)	0.0765 (8)	0.0591 (7)	0.0159 (6)	0.0000 (6)	0.0020 (6)
O4B	0.0855 (10)	0.0592 (7)	0.0587 (8)	0.0038 (6)	−0.0165 (7)	0.0077 (6)
C6B	0.0573 (10)	0.0513 (9)	0.0462 (9)	0.0001 (7)	0.0048 (8)	0.0017 (7)
C7B	0.0483 (9)	0.0521 (8)	0.0441 (8)	0.0068 (7)	0.0102 (7)	0.0045 (7)
C4B	0.0492 (9)	0.0614 (9)	0.0452 (8)	0.0050 (7)	0.0119 (7)	0.0006 (7)
C8B	0.0540 (9)	0.0444 (8)	0.0448 (8)	0.0049 (6)	0.0119 (7)	0.0017 (6)
C9B	0.0564 (10)	0.0574 (9)	0.0510 (9)	0.0078 (7)	0.0073 (8)	0.0010 (8)
C3B	0.0585 (11)	0.0549 (9)	0.0519 (9)	0.0040 (7)	0.0055 (8)	−0.0035 (7)
C11B	0.0587 (11)	0.0580 (10)	0.0537 (9)	−0.0044 (8)	0.0174 (8)	−0.0079 (8)
C10B	0.0556 (11)	0.0672 (11)	0.0557 (10)	−0.0018 (8)	0.0041 (8)	−0.0068 (8)
C1B	0.0866 (15)	0.0621 (11)	0.0799 (13)	−0.0109 (10)	0.0070 (12)	−0.0092 (10)
C2B	0.0880 (15)	0.0808 (13)	0.0711 (12)	0.0114 (11)	0.0326 (12)	−0.0039 (10)
C12B	0.0895 (16)	0.0570 (11)	0.0872 (14)	−0.0067 (10)	0.0246 (12)	−0.0075 (10)
O5A	0.0561 (7)	0.0374 (5)	0.0592 (7)	0.0039 (4)	0.0061 (5)	0.0076 (5)
O2A	0.0550 (7)	0.0492 (6)	0.0600 (7)	0.0048 (5)	0.0018 (5)	0.0103 (5)
O3A	0.0549 (7)	0.0477 (6)	0.0712 (8)	−0.0089 (5)	0.0001 (6)	0.0026 (5)
C5A	0.0480 (9)	0.0354 (7)	0.0479 (8)	0.0017 (6)	0.0090 (7)	0.0018 (6)
O1A	0.0849 (9)	0.0341 (5)	0.0708 (8)	0.0066 (5)	−0.0116 (7)	0.0015 (5)
C8A	0.0480 (9)	0.0367 (7)	0.0473 (8)	0.0040 (6)	0.0092 (7)	0.0059 (6)
C7A	0.0463 (9)	0.0392 (7)	0.0506 (8)	−0.0023 (6)	0.0073 (7)	−0.0013 (6)
C4A	0.0493 (9)	0.0410 (8)	0.0469 (8)	0.0027 (6)	0.0113 (7)	0.0024 (6)
C6A	0.0624 (11)	0.0376 (8)	0.0660 (11)	0.0023 (7)	−0.0038 (9)	0.0035 (7)

O4A	0.0940 (11)	0.0423 (6)	0.1015 (11)	−0.0076 (6)	−0.0370 (9)	0.0016 (7)
C11A	0.0673 (11)	0.0376 (7)	0.0568 (9)	0.0069 (7)	0.0209 (9)	0.0080 (7)
C9A	0.0588 (10)	0.0407 (8)	0.0623 (10)	−0.0016 (7)	0.0094 (8)	0.0012 (7)
C3A	0.0724 (12)	0.0395 (8)	0.0562 (10)	0.0010 (7)	0.0018 (9)	0.0059 (7)
C10A	0.0763 (13)	0.0325 (7)	0.0738 (11)	−0.0008 (7)	0.0185 (10)	0.0026 (7)
C2A	0.1048 (18)	0.0537 (11)	0.0846 (14)	0.0204 (10)	−0.0177 (12)	0.0118 (10)
C12A	0.0846 (14)	0.0530 (10)	0.0762 (12)	0.0200 (9)	0.0180 (11)	0.0194 (9)
C1A	0.1165 (19)	0.0727 (13)	0.0917 (15)	−0.0048 (12)	0.0447 (15)	0.0214 (12)

Geometric parameters (Å, °)

O2B—C4B	1.348 (2)	O5A—C11A	1.3542 (17)
O2B—C3B	1.433 (2)	O5A—C8A	1.3851 (18)
O5B—C11B	1.351 (2)	O2A—C4A	1.3662 (18)
O5B—C8B	1.3914 (17)	O2A—C3A	1.425 (2)
O1B—C6B	1.354 (2)	O3A—C4A	1.1985 (19)
O1B—C3B	1.428 (2)	C5A—C7A	1.358 (2)
C5B—C7B	1.358 (2)	C5A—C4A	1.452 (2)
C5B—C6B	1.460 (2)	C5A—C6A	1.467 (2)
C5B—C4B	1.467 (2)	O1A—C6A	1.3545 (19)
O3B—C4B	1.2070 (19)	O1A—C3A	1.431 (2)
O4B—C6B	1.202 (2)	C8A—C9A	1.364 (2)
C7B—C8B	1.408 (2)	C8A—C7A	1.4056 (19)
C7B—H7BA	0.9300	C7A—H7AA	0.9300
C8B—C9B	1.366 (2)	C6A—O4A	1.194 (2)
C9B—C10B	1.391 (2)	C11A—C10A	1.344 (3)
C9B—H9BA	0.9300	C11A—C12A	1.481 (2)
C3B—C2B	1.499 (3)	C9A—C10A	1.399 (2)
C3B—C1B	1.502 (2)	C9A—H9AA	0.9300
C11B—C10B	1.354 (2)	C3A—C1A	1.491 (3)
C11B—C12B	1.480 (2)	C3A—C2A	1.514 (2)
C10B—H10A	0.9300	C10A—H10B	0.9300
C1B—H1BA	0.9600	C2A—H2AA	0.9600
C1B—H1BB	0.9600	C2A—H2AB	0.9600
C1B—H1BC	0.9600	C2A—H2AC	0.9600
C2B—H2BA	0.9600	C12A—H12D	0.9600
C2B—H2BB	0.9600	C12A—H12E	0.9600
C2B—H2BC	0.9600	C12A—H12F	0.9600
C12B—H12A	0.9600	C1A—H1AA	0.9600
C12B—H12B	0.9600	C1A—H1AB	0.9600
C12B—H12C	0.9600	C1A—H1AC	0.9600
C4B—O2B—C3B	118.43 (13)	C11A—O5A—C8A	107.77 (13)
C11B—O5B—C8B	107.37 (13)	C4A—O2A—C3A	118.26 (13)
C6B—O1B—C3B	119.91 (13)	C7A—C5A—C4A	125.38 (14)
C7B—C5B—C6B	124.95 (15)	C7A—C5A—C6A	115.42 (15)
C7B—C5B—C4B	116.48 (14)	C4A—C5A—C6A	119.16 (13)
C6B—C5B—C4B	118.43 (15)	C6A—O1A—C3A	119.26 (12)
O4B—C6B—O1B	116.99 (15)	C9A—C8A—O5A	108.02 (13)
O4B—C6B—C5B	126.11 (16)	C9A—C8A—C7A	139.38 (15)

O1B—C6B—C5B	116.81 (14)	O5A—C8A—C7A	112.60 (13)
C5B—C7B—C8B	133.72 (14)	C5A—C7A—C8A	134.05 (16)
C5B—C7B—H7BA	113.1	C5A—C7A—H7AA	113.0
C8B—C7B—H7BA	113.1	C8A—C7A—H7AA	113.0
O3B—C4B—O2B	118.18 (16)	O3A—C4A—O2A	117.16 (15)
O3B—C4B—C5B	124.96 (16)	O3A—C4A—C5A	126.53 (14)
O2B—C4B—C5B	116.81 (14)	O2A—C4A—C5A	116.19 (13)
C9B—C8B—O5B	107.91 (15)	O4A—C6A—O1A	117.49 (15)
C9B—C8B—C7B	138.86 (15)	O4A—C6A—C5A	125.89 (14)
O5B—C8B—C7B	113.20 (13)	O1A—C6A—C5A	116.55 (15)
C8B—C9B—C10B	107.48 (16)	C10A—C11A—O5A	109.20 (14)
C8B—C9B—H9BA	126.3	C10A—C11A—C12A	133.71 (15)
C10B—C9B—H9BA	126.3	O5A—C11A—C12A	117.09 (16)
O1B—C3B—O2B	110.13 (13)	C8A—C9A—C10A	106.92 (16)
O1B—C3B—C2B	110.27 (16)	C8A—C9A—H9AA	126.5
O2B—C3B—C2B	109.85 (15)	C10A—C9A—H9AA	126.5
O1B—C3B—C1B	105.73 (14)	O2A—C3A—O1A	109.72 (13)
O2B—C3B—C1B	106.35 (16)	O2A—C3A—C1A	110.21 (15)
C2B—C3B—C1B	114.36 (16)	O1A—C3A—C1A	110.72 (17)
O5B—C11B—C10B	109.69 (15)	O2A—C3A—C2A	106.18 (16)
O5B—C11B—C12B	116.80 (16)	O1A—C3A—C2A	105.14 (14)
C10B—C11B—C12B	133.51 (18)	C1A—C3A—C2A	114.63 (17)
C11B—C10B—C9B	107.55 (17)	C11A—C10A—C9A	108.08 (15)
C11B—C10B—H10A	126.2	C11A—C10A—H10B	126.0
C9B—C10B—H10A	126.2	C9A—C10A—H10B	126.0
C3B—C1B—H1BA	109.5	C3A—C2A—H2AA	109.5
C3B—C1B—H1BB	109.5	C3A—C2A—H2AB	109.5
H1BA—C1B—H1BB	109.5	H2AA—C2A—H2AB	109.5
C3B—C1B—H1BC	109.5	C3A—C2A—H2AC	109.5
H1BA—C1B—H1BC	109.5	H2AA—C2A—H2AC	109.5
H1BB—C1B—H1BC	109.5	H2AB—C2A—H2AC	109.5
C3B—C2B—H2BA	109.5	C11A—C12A—H12D	109.5
C3B—C2B—H2BB	109.5	C11A—C12A—H12E	109.5
H2BA—C2B—H2BB	109.5	H12D—C12A—H12E	109.5
C3B—C2B—H2BC	109.5	C11A—C12A—H12F	109.5
H2BA—C2B—H2BC	109.5	H12D—C12A—H12F	109.5
H2BB—C2B—H2BC	109.5	H12E—C12A—H12F	109.5
C11B—C12B—H12A	109.5	C3A—C1A—H1AA	109.5
C11B—C12B—H12B	109.5	C3A—C1A—H1AB	109.5
H12A—C12B—H12B	109.5	H1AA—C1A—H1AB	109.5
C11B—C12B—H12C	109.5	C3A—C1A—H1AC	109.5
H12A—C12B—H12C	109.5	H1AA—C1A—H1AC	109.5
H12B—C12B—H12C	109.5	H1AB—C1A—H1AC	109.5

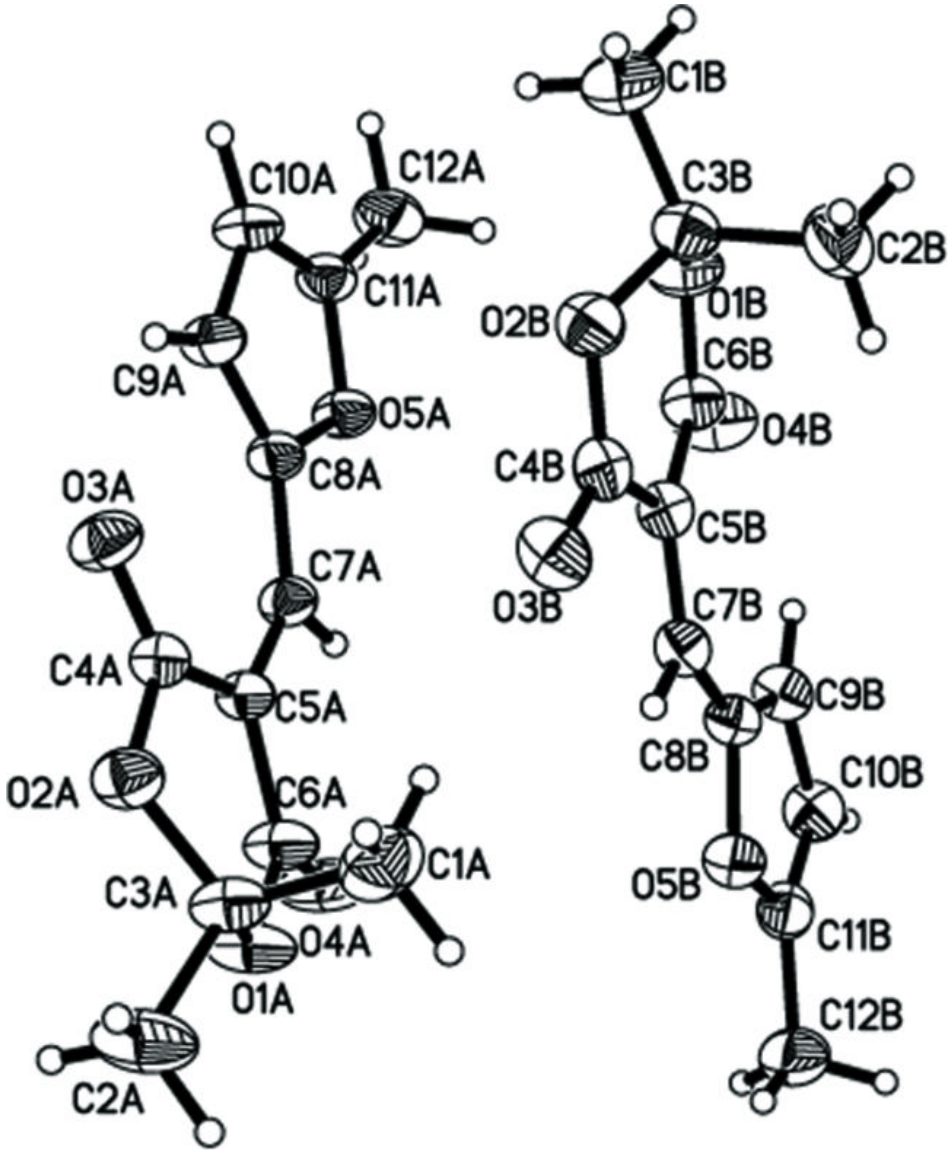
Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
C1B—H1BC···O3B^{i}	0.96	2.48	3.434 (3)	170
C10B—H10A···O4A^{ii}	0.93	2.60	3.448 (3)	153

C10A—H10B···O1A^{iii}	0.93	2.44	3.343 (2)	163
C2A—H2AB···O2A^{iv}	0.96	2.57	3.524 (3)	170
C1A—H1AA···O3B	0.96	2.55	3.496 (3)	170

Symmetry codes: (i) −*x*, −*y*+1, −*z*+1; (ii) −*x*+1, −*y*+2, −*z*+2; (iii) *x*, *y*−1, *z*; (iv) −*x*−1, −*y*+2, −*z*+1.

Fig. 1


C1B
C3B
C2B
C10A
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C11A
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O2B
C6B
C9A
O5A
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C8A
O3A
C5B
C7A
O3B
C7B
C4A
C5A
C9B
C8B
C10B
O2A
C6A
C1A
O5B
C3A
O4A
C11B
O1A
C12B
C2A